DITERPENOIDS FROM LIVERWORTS*

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Dedicated to Professor F. Šantavý on the occasion of his 60th birthday.

Six diterpenoids were isolated from the light petroleum extracts of three liverworts, *Jungermannia sphaerocarpa* Hook. [syn. *Solenostoma* s. (Hook.) Steph.], *Pellia endiviifolia* (DICKS.) DUM., and *Mylia anomala* (Hook.) S. GRAY.

In the course of a systematic investigation of liverworts we mainly concentrated on their sesquiterpenic components¹⁻⁵. However, during these studies we did not neglect other substances either, which were the major components⁶⁻⁸ of these as yet little studied plants. In this paper we describe substances of diterpenoid character which are — as is evident from our extensive study — represented in this group of plants in smaller quantities than sesquiterpenes. A similar line of research was also followed by other authors studying diterpenoids which were isolated from the liverworts of Jungermanniaceae^{9,10}, or Antheliaceae¹¹ and Marchantiaceae¹² families.

EXPERIMENTAL AND RESULTS

The melting points were determined on a Kofler block and they are not corrected. The IR spectra were measured in chloroform by means of a Zeiss UR 10 spectrophotometer (Jena), the UV spectra on a CF 4 apparatus (Optica Milano), and the mass spectra on an AEI MS 902 spectrometer. The PMR spectra in deuteriochloroform were recorded with a Varian HA-100 apparatus using tetramethylsilane as internal standard. Silica gel for column chromatography was prepared according to Pitra and Štěrba¹³ and was deactivated with 10% of water. The neutral alumina used (Laborchemie Apolda) had activity grade V according to Brockmann¹⁴.

Jungermannia sphaerocarpa Hook. [syn. Solenostoma s. (Hook.) Steph.], (Jungermanniaceae)

Origin. Valley of Černá Nisa near Rudolfov (Liberec), Czechoslovakia. The liverwort (3600 g) was dried at room temperature, ground, and extracted with light petroleum (b.p. $40-60^{\circ}$ C);

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gel¹³ using a mixture of benzene and diethyl ether (5–10%) for elution. A mixture of crystalline compounds was eluted, which was separated by chromatography on alumina (act. V., neutral). Elution with light petroleum gave component A (356 mg), m.p. $110-111^{\circ}$ C (benzene), R_F 0·63 (benzene-diethyl ether 7: 3 on silica gel G). IR (6% in chloroform): 897, 928, 976, 989, 1018, 1044, 1052, 1092, 1115, 1132, 1195, 1229 (ester), 1270, 1290, 1372, 1412, 1443, 1462, 1656 (C=C), 1746 (ester), 2845, 2865, 2910, 2930, 3580 (OH) cm⁻¹; MS: M⁺ 346, C₂₂H₃₄O₃; PMR (CDCl₃):

0·80 s, 0·86 s, 0·91 s, 3 × 3 H, 3 × CH₃—C—; 2·19 s, 3 H, CH₃—C=O; 2·5–2·7 mt, 2 H; 3·81 mt, W = 24, 1 H, CH—O—; 4·94 mt, 5·14 mt, 2 × 1 H, C=CH₂, J_{gem} = 0; 5·30 t, 1 H, CHOCOCH₃, shifts in δ -scale.

from the concentrated extract a residue (16 g) was obtained, which was chromatographed on silica

Elution with light petroleum containing 5% of benzene has given substance B (42 mg), m.p. $131-132^{\circ}\mathrm{C}$ (benzene), R_F 0·40 (light petroleum–acetone 8 : 2; alumina Woelm neutral for TLC), or R_F 0·59 (benzene–diethyl ether 7 : 3; silica gel G). IR (6% in chloroform); 795, 940, 977, 1029, 1052, 1060, 1114, 1180, 1186, 1235, 1260, 1342, 1370, 1390, 1450, 1460, 1645 (C=C), 1722 (C=O), 2780, 2935, 3000, 3080, 3480 (OH), 3595 (OH), 3660 cm⁻¹; UV (ethanol): 238 nm (log ε 3·81); MS: M⁺ 302, $C_{20}H_{30}O_2$; PMR (CDCl₃): 0·82 s, 0·90 s, 1·02 s, 3 × 3 H, 3 × CH₃— C-; 2·40 d, $J=12\cdot3$, 1 H; 3·04 mt, 1 H, allylic H; 4·05 d, J=5, 1 H, CH—O—; 5·25 mt, 5·86 mt, 2 × 1 H, C=CH₂, $J_{\mathrm{gem}}=1\cdot0$, shifts in δ -scale.

Elution with light petroleum containing 30% of benzene gave substance C (295 mg), m.p. $192-194^{\circ}\text{C}$ (benzene), R_F 0·43 (benzene–diethyl ether 7 : 3; silica gel G). IR (6% in chloroform): 831, 850, 866, 894, 923, 948, 963, 992, 1014, 1028, 1040, 1056, 1068, 1077, 1091, 1111, 1133, 1193, 1239, 1242, 1275, 1371, 1390, 1442, 1462, 1580, 1658 (C=C), 2674, 2845, 2870, 2930, 2955, 3070 (C=C), 3430 (OH), 3590 (OH) cm⁻¹; MS: M⁺ 304, C₂₀H₃₂O₂; PMR (CDCl₃): 0·82 s, 0·88 s, 0·95 s, 3 × 3 H, 3 × CH₃—C—; 2·62 mt, 1 H, allylic proton; 3·58 mt, W = 8, 1 H, CH—O—; 4·01 d, J=5, 1 H, CH—O—; 5·02 bs + 5·12 bs, C=CH₂, shifts in δ -scale.

From the characteristics of the above substances it follows that they are of diterpenic character. In the case of substance A it is a hydroxy acetate, substance B is a hydroxy ketone, and substance C is a diol.*

Pellia endiviifolia (DICKS.) DUM. (Pelliaceae)

Origin. Srbsko (Karlštejn), Czechoslovakia. The liverwort (1300 g) was dried at room temperature, then ground and extracted with light petroleum (b.p. $40-60^{\circ}$ C); the concentrate (5 g) was chromatographed on silica gel² deactivated with 15% of water, using a 1:1 mixture of light petroleum and benzene for elution. A substance (20 mg) was obtained, m.p. 120°C (light petroleum), R_F 0·35 (benzene-diethyl ether 95:5; silica gel G). IR (5% in tetrachloromethane): 865,

^{*} From the liverwort Solenostoma triste (NEES) K. MÜLL. (= Jungermannia atrovirens Dum.) Connolly and Thornton isolated recently four diterpenoids and assigned then the structures derived from 11β-hydroxylated ent-kaurene^{1.5}.

924, 955, 980, 1070, 1109, 1130, 1174, 1192, 1208, 1255, 1300, 1344, 1365, 1386, 1422, 1466, 1636 (C=C), 1689 (C=C-C=O), 1714 (-CHO), 2730 cm⁻¹; MS: M⁺ 302, $C_{20}H_{30}O_2$; PMR (CDCl₃): 0.95 s, 0.97 s, 2 × 3 H, 2 × CH₃—C—; 1.58 bs, 1.67 bs, 2 × 3 H, 2 × CH₃—C=C; 2.46 mt, 2 H, 2.85 mt, 1 H, 5.06 mt, 1 H, -CH=C—; 7.12 mt, 1 H, -CH=C—; 9.47 s, 1 H, -CH=C, 9.47 s, 1 H, -CH=C), shifts in δ -scale.

From the characteristic data of the above diterpenoid it follows that it is a dialdehyde.

Mylia anomala (HOOK.) s. GRAY (Jungermanniaceae)

Origin: Mount Sv. Sebastian (Chomutov), Czechoslovakia. The liverwort (130 g) was dried at room temperature, then ground and extracted with light petroleum (b.p. $40-60^{\circ}$ C); from the concentrate of the light petroleum extract a crystalline mixture separated which, after filtration of the mother liquors was separated chromatographically on silica gel² deactivated by 10% of water. Elution with benzene gave substance A (100 mg), m.p. 153° C (benzene-diethyl ether), R_F 0.46 (benzene; silica gel G). IR (6% in chloroform): 858, 941, 970, 1015, 1035, 1075, 1100, 1122, 1162, 1306, 1321, 1365, 1386, 1455, 1464, 3 060 (cyclopropane ring), 3 620 (OH) cm⁻¹; MS: M⁺ 306, C₂₀H₃₄O₂; PMR (CDCl₃): 0.18 t, $J_{\text{gem}} = 4.5$, 1 H, 0.43 dd, $J_{\text{gem}} = 4.5$, $J_{\text{vic}} = 8.5$, 1 H, two geminal cyclopropane protons; 0.76 s, 1.03 s, 1.19 s, 3 × 3 H, 3 × CH₃—C—; 0.86 d, J = 6.7, 0.91 d, J = 6.7, 2 × 3 H, 2 × CH₃—CH—; 3.65 mt, 2 H, 2 × CH—O—H, shifts in δ -scale.

Elution with benzene containing 20% of diethyl ether gave substance B (10 mg), m.p. 157°C (benzene), R_F 0·18 (benzene; silica gel G). IR (6% in chloroform): 857, 904, 915, 950, 975, 990, 1035, 1070, 1092, 1118, 1168, 1305, 1333, 1385, 1412, 1462, 1636 (C—C), 3090 (C—C), 3620 (OH) cm⁻¹; MS: M⁺ 322, C₂₀H₃₄O₃; PMR (CDCl₃): 0·91 s, 0·95 s, 1·25 s, 1·30 s, 4 × 3 H, 4 × CH₃—C—; 3·74 mt, 1 H, 3·89 mt, 1 H, 2 × CH—O—; 4·82 dd, J_{cis} = 11, J_{gem} = 1·5, 1 H, 4·88 dd, J_{trans} = 17, J_{gem} = 1·5, 1 H, 5·76 dd, J_{trans} = 17, J_{cis} = 11, 1 H, CH₂=CH—C.

From the data for the mentioned diterpenoids from M. anomala it follows that substance A is a diol, while substance B is a triol.

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